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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds

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7 Chemical Formulas and Chemical Compounds

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Title: Modern Chemistry Chapter 7 Chemical Formulas 1 Modern Chemistry Chapter 7 Chemical Formulas Chemical Compounds . A chemical formula indicates the kind and relative number of atoms in a chemical compound. C_8H_{18} (octane) has 8 carbon and 18 hydrogen atoms. 2 Forming Ionic Compounds. Compounds that have the elements held together by

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Answer the following questions in the space provided Modern chemistry chapter 7 review answers section 4. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

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Modern Chemistry . Name Section Quiz, continued Class Date 6. How many atoms are present in one formula unit of barium acetate, c. 15 d. 16 7. What is the formula for the compound dinitrogen tetroxide? a. d. NO 8. The term acid usually refers to a. a solution of the acid compound in water.

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Modern Chemistry • CHAPTER 7 HOMEWORK 7-2 ... CHAPTER 7 • HOMEWORK 7-2 VOCABULARY 1. d
2. c 3. e 4. a 5. g 6. h 7. b 8. f GRAPHIC ORGANIZER Name of Compound or Ion Formula of
Compound Copper(II) ion Cu^{2+} Ammonium ion NH_4^+ Bismuth(III) bromide BiBr_3 Stannous
fluoride SnF_2 Magnesium fluoride MgF_2 Iron(III) cyanide $\text{Fe}(\text{CN})_3$

HOMEWORK 7-2

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